Association of sinking organic matter with various types of mineral ballast in the deep sea: Implications for the rain ratio

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[1] We compiled and standardized sediment trap data below 1000 m depth from 52 locations around the globe to infer the implications of the Armstrong et al. [2002] "ballast" model to the ratio of organic carbon to calcium carbonate in the deep sea (the rain ratio). We distinguished three forms of mineral ballast: calcium carbonate, opal, and lithogenic material. We concur with Armstrong et al. [2002] that organic carbon sinking fluxes correlate tightly with mineral fluxes. Based on the correlations seen in the trap data, we conclude that most of the organic carbon rain in the deep sea is carried by calcium carbonate, because it is denser than opal and more abundant than terrigenous material. This analysis explains the constancy of the organic carbon to calcium carbonate rain ratio in the deep sea today, and argues against large changes in the mean value of this ratio in the past. However, sediment trap data show variability in the ratio in areas of high relative calcium carbonate export (mass CaCO₃/mass ratio > 0.4), unexplainable by the model, leaving open the possibility of regional variations in the rain ratio in the past. INDEX TERMS: 1615 Global Change: Biogeochemical processes (4805); 1635 Global Change: Oceans (4203); 4863 Oceanography: Biological and Chemical: Sedimentation; KEYWORDS: organic carbon flux, sediment traps, marine carbon cycle, oceanography, rain ratio

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1. Introduction

[2] Model studies have proposed an increase in ocean pH as a mechanism to explain the decrease in atmospheric pCO₂ during glacial time [*Archer et al.*, 2000]. The increase in pH of the ocean could be driven by an increase in the weathering minus shallow water CaCO₃ deposition [*Opdyke and Walker*, 1992], but the lysocline signature of an increase in deep sea CaCO₃ burial during the last glacial maximum is not seen [*Archer and Maier-Reimer*, 1994; *Catubig et al.*, 1998]. The pH of the ocean may also be sensitive to the ratio of particulate organic carbon (POC) to CaCO₃ or particulate inorganic carbon (PIC) raining to the deep sea (the POC/PIC rain ratio), which may have a smaller impact on the distribution of CaCO₃ on the seafloor [*Archer et al.*, 2000].

[3] Export of POC into the deep ocean is thought to be driven primarily by growth and sedimentation of diatoms following nutrient injection events due to winter convection or upwelling [*Peinert et al.*, 1989; *Smetacek et al.*, 1990; *Ittekkot et al.*, 1991; *Honjo*, 1996; *Takahashi et al.*, 2000]. CaCO₃ production, mediated primarily by coccoli-

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thophores, is thought to increase when diatom growth is limited by availability of silica or iron [Margalef, 1978; Tyrell and Taylor, 1996]. However, if organic carbon export to the deep sea were associated primarily with diatom production at the sea surface, we would expect regional variability in the rain ratio that is not observed in trap data, and could not be tolerated by models of CaCO₃ preservation on the seafloor [Archer, 1996]. Armstrong et al. [2002] and François et al. [2002] propose that organic carbon export is determined by the presence of ballast minerals (opal, calcium carbonate and lithogenic material). The model of Armstrong et al. [2002] partitions POC in sinking matter into two fractions: a ballast-associated fraction with remineralization scale similar to the dissolution scales of the minerals, and a "free" POC fraction which decays in the top kilometer or so. Either the ballast increases the sinking speed of the organic matter, as implied by the term ballast, or it could provide some physical protection from degradation through adsorption or other interaction between organic matter and minerals. Armstrong et al. [2002] found significant geographical differences in the POC fraction associated with the mineral fraction and hypothesized that such differences might be related to differences in composition of the ballast material. The purpose of this study is to synthesize a global long-term sediment trap data set to investigate the relationship between mineral fractions (opal, calcium carbonate

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and lithogenic material) and the transfer of POC to the deep ocean, based on the model of *Armstrong et al.* [2002].

2. Methods

2.1. Data

[4] We compiled time series and composite annual flux data from sediment trap experiments at 52 different locations with a minimum duration of 10 months and at depths ranging from 1000 to 4833 m (Figure 1, Table 1).

[5] Biogenic silica fluxes were determined using different methods depending on authors. The most widespread methods are modifications of the sodium carbonate/sodium hydroxide sequential leaching method. Some authors, however, use a normative method based on aluminum and total silica (biogenic and lithogenic) measurements and assuming a Si/Al ratio varying from 3 to 3.5 for lithogenic component of the silica fluxes. In this study we standardized estimates on an Al/Si ratio of 3 when no biogenic silica determination was carried out. Directly measured biogenic silica fluxes were corrected by standardizing the water fraction using a water/silicon weight ratio of 0.256 (10% water, [Mortlock and Froelich, 1989]). For trap experiments where only the combustible fraction was measured, POC has been estimated using a particulate organic matter (POM) to POC ratio of 2.199 (Figure 2). This ratio was obtained though regression analysis of POC versus POM from trap experiments in the San Pedro Basin [Thunell et al., 1994] and from suspended matter analysis in the Southern Ocean [Suess and Ungerer, 1981]. POM values were estimated from the measured combustible fraction corrected in both data sets for water content of opal [Tsunogai and Noriki, 1987] using a water/silicon weight ratio of 0.256 [Mortlock and Froelich, 1989]. The lithogenic component in the traps was estimated as total mass less calcium carbonate (CaCO₃), opal (SiO₂0.4H₂O), and organic matter (POC*2.199 or the combustible fraction less water content of measured opal flux). For a few trap experiments, data on the fraction of POC and opal that was remineralized or dissolved in the cups was available [Honjo and Manganini, 1993; Honjo et al., 1995] and was added to the particulate fluxes.

[6] Errors for the annual flux data compiled in this study were estimated whenever analytical errors were given by the authors. Splitting errors were accounted for using the given values, or assumed to be 6%, the maximum value found in the literature [von Bodungen, 1991; Honjo and Manganini, 1993; Honjo et al., 1995, 1999, 2000; Nodder and Northcote, 2001]. A few studies give estimates of the fraction of material dissolved or remineralized in the sampling cups (POC and/or silica) together with analytical errors. In such cases analytical error of the dissolved fraction was included in the error calculations. Error estimates could be obtained for traps at 21 different locations (Table 1).

2.2. Mineral-Associated POC Flux Model

[7] The formulation of *Armstrong et al.* [2002] is a mechanistic model that was developed as a best fit to the vertical variations in POC with depth as measured in sedi-

ment traps. The model assumes that part of the export POC is associated with a ballast fraction and is proportional to this fraction. The excess POC is considered the free fraction:

$$\mathbf{F}_{\mathbf{p}} = \mathbf{F}_{\mathbf{e}} + f.\mathbf{F}_{\mathbf{b}},\tag{1}$$

where F_p is the total export POC mass flux, F_e is the mass flux of the free POC, F_b is the total ballast mass flux, and *f* is a proportionality factor, essentially the POC carrying capacity or coefficient of the ballast. The different components of the POC sinking flux decrease with depth according to exponential or power law functions

$$\mathbf{F}(\mathbf{z}) = \mathbf{F}_{\mathbf{e}} \cdot \mathbf{f}(\mathbf{z}, \partial_{\mathbf{e}}) + f \cdot \mathbf{F}_{\mathbf{b}} \cdot \mathbf{f}(\mathbf{z}, \partial_{\mathbf{b}}), \tag{2}$$

where F(z) is the total POC flux at depth z, f(z, ∂_e) and f(z, ∂_b) are decay functions for free and ballasted POC.

[8] Here we extend the application of the model by distinguishing the different forms of mineral ballast (CaCO₃, opal (SiO₂ 0.4 H₂O), and lithogenic terrigenous material (clays)) to predict large scale variability of the POC/PIC rain ratio in the deep sea as suggested by *Armstrong et al.* [2002]. Based on the approximately 500 meter depth decay scale for "free" POC from *Armstrong et al.* [2002], we first assume that below 1000 m the free POC flux is negligible. The model reduces to:

$$F(z) = f_{ca}.F_{ca}(z) + f_{o}.F_{o}(z) + f_{1}.F_{1}(z),$$
(3)

where $F_o(z)$, $F_{ca}(z)$, and $F_1(z)$ are the mass fluxes of CaCO₃, opal, and lithogenic material at depth z. The carrying coefficients *f* can be determined by multiple linear regression of trap data below 1000 m depth.

3. Results

[9] Scatterplots of POC versus mineral constituents of annual fluxes for trap experiments below 1000 m depth are given in Figure 3 and indicate a tendency toward increasing POC with mineral flux (CaCO₃, opal, and lithogenic material). Multiple correlation analysis of POC fluxes as a function of mineral fraction (in mass units) for different trap data sets is highly significant with a correlation coefficient varying from 0.956-0.985 (Table 2). The carrying coefficients are highest for CaCO₃ and lowest for opal mass fluxes. Using different data sets for the analysis does not change significantly carrying coefficients for the different mineral fractions, however, the carrying coefficients tend to decrease when using deep trap data, suggesting that a small fraction of the POC transported to 1000 m is remineralized before reaching the sediments. Overall carrying coefficients determined using trap experiments below 2000 m seem to show the least uncertainty. If the data set is reduced to sediment trap experiments carried out in open ocean areas far from influence of "exogenous" sources such as resuspended sediments from continental shelves or ice-rafted debris ("selected" sediment trap experiments, Figure 1), the coefficients for CaCO₃ and opal do not change much. The coefficient for the lithogenic fraction shows the highest





Table 1. Location, Water Depth, Trap Depth, Period Sampled, Duration, Codes and References of the Annual Trap Experiment DataUsed in This Study

		Water	Trap				
Latituda	Longitudo	Depth,	Depth,	Someting David	Duration,	Cada	Deference
Latitude	Longitude	m	m	Sampling Period	days	Code	Reference
53°32.0'N	176°56.2′W	3783	3193	7/8/90-17/7/91	344	AB1	Takahashi et al. [2000]
53°31.0 N	177°05.0 W	3790	3200	$\frac{1}{8}\frac{91-31}{192}$	359	AB2	Takahashi et al. [2000]
53°30 4'N	177 04.2 W	3788	3199	2/8/95 - 7/0/94 4/8/94 - 25/7/95	309	AD4 AB5	Takahashi et al. [2000] Takahashi et al. [2000]
63°09.0′S	169°54 0'W	2885	1031	$\frac{1}{1/97} = \frac{23}{1/98}$	425	ACCa	Honio et al. $[2000]$
60°17.0′S	170°03.0′W	3957	1003	1/1/97 - 27/1/98	425	APF ^a	Honjo et al. [2000]
17°12.0′N-	59°36.0′E-	3465-	1857-	11/11/94-24/12/95	391	ASm ^a	Honjo et al. [1999]
17°13.0′N	59°36.0′E	3477	1882			"	Honjo et al. [1999]
17°12.0′N-	59°36.0′E-	3465-	2871-	11/11/94-24/12/95	391	ASm ^a	Honjo et al. [1999]
17°13.0′N	59°36.0′E	3447	2991			"	Honjo et al. [1999]
17°24.0′N	58°48.0′E-	3642-	3141-	11/11/94-24/12/95	391	ASn ^a	Honjo et al. [1999]
10000 0/01	58°48.0'E	3655	3159		201		Honjo et al. [1999]
10°00.0 N	65°00.0 E	4411	2363	28/11/94-24/12/95	391	AS0"	Honjo et al. [1999]
10 00.0 N	65 00.0 E	4411 2074	2220	28/11/94-24/12/95	391	AS0	Honjo et al. $[1999]$
13 20.0 N- 12°59 0'N	61°30.0'E	39/4-	2229-		574	ASS "	Honjo et al. $[1999]$ Honjo et al. $[1999]$
15°20 0'N-	61°30 0′E-	3974-	3478-	11/11/94-24/12/95	391	ASs ^a	Honjo et al. [1999]
12°59.0′N	61°30.0′E	3983	3489		571	"	Honjo et al. $[1999]$
75°51.0′N	11°28.0′E	2123	1650	12/07/84-10/07/85	363	BI	Honjo et al. [1988]
1°00.0′N	139°00.0′W	4470	1095	12/12/82-25/2/84	428	C3	Dymond and Collier [1988]
1°00.0′N	139°00.0′W	4470	1895	12/12/82-25/2/84	428	C3	Dymond and Collier [1988]
1°00.0′N	139°00.0′W	4470	3495	12/12/82-25/2/84	428	C3	Dymond and Collier [1988]
1°00.0′N	139°00.0′W	4470	1083	25/2/84-22/2/85	363	C4	Dymond and Collier [1988]
1°00.0′N	139°00.0′W	4470	1883	25/2/84-22/2/85	363	C4	Dymond and Collier [1988]
1°00.0'N	139°00.0'W	4470	2908	25/2/84-22/2/85	363	C4	Dymond and Collier [1988]
14°29.0'N-	$64^{\circ}46.0^{\circ}E^{\circ}$	3912-	2894-	20/11/80-2/5/8/	>305	CASI"	Haake et al. [1993]; Ramaswamy and Nair [1994]
14 28.0 N- 14°33 0'N	64 40.0 E-	3890-	2913-	12/5/8/-21/10/8/ 22/11/87 31/10/88		"	Haake et al. [1993]; Ramaswamy and Nair [1994] Haake et al. [1993]; Ramaswamy and Nair [1994]
14°33 0′N-	64°47.0'E	3910-	2933-	22/11/87 = 31/10/88 22/11/87 = 31/10/88	>305	CAS2 ^a	Haake et al. [1993]: Ramaswamy and Nair [1994]
14°36.0′N	"	3906	3109	20/11/88-24/2/89	- 505	"	Haake et al. [1993]: Ramaswamy and Nair [1994]
21°08.7′N	20°41.2′W	4092	3502	15/3/89-24/3/90	374	CB1 ^a	Wefer and Fischer [1993]
21°08.0′N-	20°40.0′W	4094-	3557-	1/1/91-19/11/91	298	CB2 ^a	Fischer and Wefer [1996]
21°09.0′N	20°41.0′W	4108	3562			"	Fischer and Wefer [1996]
20°55.3′N	19°44.5′W	3646	2195	22/3/88-8/3/89	351	CBa ^a	Wefer and Fischer [1993]
13°09.0′N	84°22.0′E-	3259-	2282-	28/10/87-28/2/88	312	CBB1	Ittekkot et al. [1991]; Schäfer et al. [1996]
"	84°21.0′E	3263	2227	4/1/88-10/6/88			Ittekkot et al. [1991]; Schäfer et al. [1996]
13°09.0'N	84°21.0'E	3263	2286	2/11/88-19/10/89	351	CBB2	Ittekkot et al. [1991]; Schäfer et al. [1996]
15 09.0 N 30°01 1/S	64 20.0 E 73°11 0'W	5207 4360	2202	$\frac{3}{12}\frac{90-20}{10}\frac{91}{91}$	340	СВВЗ	Hebbeln et al. [2000]
30°00 3′S	73°10 3′W	4360	2323		540	"	Hebbeln et al. [2000] Hebbeln et al. [2000]
15°28.0′N-	68°45.0′E-	3782-	2764-	20/11/86-2/5/87	>305	EAS ^a	Haake et al. [1993]: Ramaswamv and Nair [1994]
15°26.0′N-	68°43.0′E-	3770-	2763-	12/5/87-21/10/87		"	Haake et al. [1993]; Ramaswamy and Nair [1994]
15°32.0′N	68°45.0′E-	3782	2778	22/11/87-31/10/88		"	Haake et al. [1993]; Ramaswamy and Nair [1994]
4°07.5′N	136°16.6′E	4888	1769	4/6/91-15/4/92	317	EC^{a}	Kawahata et al. [2000]
4°07.5′N	136°16.6′E	4888	4574	4/6/91-15/4/92	317	EC^{a}	Kawahata et al. [2000]
5°00.6′N	138°49.8′E	4130	1130	21/11/88-16/12/89	390	ECC	Kempe and Knaack [1996]
5°00.6'N	138°49.8'E	4130	3130	21/11/88-16/12/89	390	ECC ED1 ^a	Kempe and Knaack [1996]
9 00.0 N 9°00 0'N	139 39.0 W	5100	2150	2/2/92 - 7/1/93 2/2/92 - 7/1/93	357	EPI FD1 ^a	Honjo et al. $[1995]$ Honjo et al. $[1995]$
5°01.0'N	139°47 0'W	4493	1191	2/2/92 = 7/1/93 2/2/92 = 7/1/93	357	EP2 ^a	Honjo et al. $[1995]$
5°01.0′N	139°47.0′W	4493	2091	2/2/92 - 7/1/93	357	EP2 ^a	Honjo et al. $[1995]$
5°01.0′N	139°47.0′W	4493	3793	2/2/92 - 7/1/93	357	EP2 ^a	Honjo et al. [1995]
2°00.0′N	140°08.0′W	4397	2203	2/2/92-7/1/93	357	EP3 ^a	Honjo et al. [1995]
0°04.0′N	139°45.0′W	4358	2284	2/2/92-7/1/93	357	EP4 ^a	Honjo et al. [1995]
0°04.0′N	139°45.0′W	4358	3618	2/2/92-7/1/93	357	EP4 ^a	Honjo et al. [1995]
1°57.0′S	139°45.0′W	4293	3593	2/2/92-7/1/93	357	EP5 ^a	Honjo et al. [1995]
4°57.0′S	139°44.0′W	4198	2099	2/2/92-7/1/93	357	EP6 ^a	Honjo et al. [1995]
4°57.0'S	139°44.0 W	4198	2209	2/2/92 - 7/1/93	357	EP6"	Honjo et al. [1995]
4 57.05 11°58 0'8	139 44.0 W	4198	2310	2/2/92 - 7/1/93	357	EPO ED7 ^a	Honjo et al. [1995] Honio et al. [1995]
11°58.0'S	135°02.0 W	4294 4294	3504	2/2/92 = 7/1/93 2/2/92 = 7/1/93	357	EP/ EP7 ^a	Honjo et al. $[1995]$
78°52.0′N	1°22.0′E	ng	2440	20/8/84 - 15/8/85	360	FS	Honjo [1990]
1°47.5′N	11°07.6′W	4481	3921	1/3/89-25/2/90	361	GB1 ^a	Wefer and Fischer [1993]
1°48.0′N	11°08.0′W	4481-	3921-	1/1/90-31/12/90	327	GB2 ^a	Fischer and Wefer [1996]
1°47.0′N	"	4522	3965			"	Fischer and Wefer [1996]
75°35.0′N	6°43.0′W	ng	2871	4/8/85-3/8/86	364	GRn	Honjo [1990]
54°32.0′N	21°04.0′W	2979	2200	10/6/92-12/5/93	336	L1	Kuss and Kremling [1999]

Table 1. (continued)

		Water Depth,	Trap Depth,		Duration,		
Latitude	Longitude	m	m	Sampling Period	days	Code	Reference
54°32.0′N	21°04.0′W	2979	2880	10/6/92-12/5/93	336	L1	Kuss and Kremling [1999]
33°09.0'N	21°59.0'W	5303	4000	20/9/93 - 1/9/94 10/6/02 - 26/5/02	346	LA	Kuss and Kremling [1999]
47 48.0 N 47°48 0'N	19 47.0 W $10^{\circ}47.0'W$	4574	2030	10/6/92 - 20/3/93 10/6/02 - 26/5/03	351	LDI I B1	Kuss and Kremling [1999]
47°48.0 N 47°48.0'N	19°47.0 W	4574	3530	10/6/92 - 26/5/93	351	LBI I R1	Kuss and Kremling [1999]
69°30 0'N	10°00 0′E	no	2760	nσ	>305	LOB	Honio [1990]
64°55.0′S	2°30.0′W	5000	4456	20/1/87 - 20/11/87	304	MR	Wefer et al. $[1990]$
47°42.9′N-	20°52.5′W-	4418-	1018-	3/4/89-26/9/89	358	NABN	Honjo and Manganini [1993]
47°43.6′N	20°51.5′W	4451	1202	16/10/89-16/4/90		"	Honjo and Manganini [1993]
47°42.9′N-	20°52.5′W-	4418-	3718-	3/4/89-26/9/89	358	NABN	Honjo and Manganini [1993]
47°43.6′N	20°51.5′W	4451	3749	16/10/89-16/4/90		"	Honjo and Manganini [1993]
33°49.3′N-	21°00.5′W-	5261-	2067-	3/4/89-26/9/89	358	NABS	Honjo and Manganini [1993]
33°48.4′N	21°02.2′W	5083	1894	16/10/89-16/4/90	2.50	"	Honjo and Manganini [1993]
33°49.3'N-	21°00.5'W-	5261-	4564-	3/4/89-26/9/89	358	NABS	Honjo and Manganini [1993]
33°48.4'N	21°02.2 W	5083	4391	16/10/89 - 16/4/90 18/4/80 - 17/0/00	502	NAm	Honjo and Manganini [1993]
4/ 30.0 N 24°33 0'N	19 50.0 W	4440	3870	16/4/89 - 17/9/90 14/10/00 - 27/0/01	302	NAn	lickells et al. [1996]
17°26 0'N	22 30.0 W 89°35 0′F-	2263-	1727-	$\frac{14}{10}\frac{30-2}{39}$	312	NRn	Ittekkot et al. [1990] Ittekkot et al. [1991]: Schäfer et al. [1996]
17°27 0'N	89°36 0′E	2203-	1790	1/4/88 - 6/10/88	512	"	Ittekkot et al. [1991]; Schäfer et al. [1996]
17°27.0′N	89°36.0′E	2265	2029	2/11/88 - 19/10/89	351	NBn	Ittekkot et al. [1991]; Schäfer et al. [1996]
15°31.0′N	89°13.0′E	2706	2120	5/12/90-26/10/91	325	NBs	Ittekkot et al. [1991]; Schäfer et al. [1996]
12°01.0′N	134°17.2′E	5300	4300	21/11/88-16/12/89	390	NEC	Kempe and Knaack [1996]
31°50.0′N	64°30.0′W	± 4500	3200	14/1/88-30/1/89	372	OFP4	Conte et al. [2001]
49°11.2′N	12°49.2′W	ng	1050	1/07/93-3/9/94	368	Ome	Wollast and Chou [1998]; Antia et al [1999]
49°05.0′N	13°25.8′W	ng	1440	1/07/93-3/9/94	368	Omw	Wollast and Chou [1998]; Antia et al [1999]
49°05.0′N	13°25.8′W	ng	3260	1/07/93-3/9/94	368	Omw	Wollast and Chou [1998]; Antia et al [1999]
50°00.0'N	145°00.0'W	4240	3800	23/9/82-23/4/84	365	OPI	Wong et. al. [1999]
50°00.0'N	145°00.0 W	4240	1000	20/11/84 - 23/4/86	360	OP3	Wong et. al. [1999]
50°00.0'N	145~00.0~W $145^{\circ}00.0'W$	4240	3800	20/11/84 - 25/4/80 10/10/86 6/5/88	2500	OP3	Wong et. al. $[1999]$ Wong et. al. $[1000]$
50°00.0'N	145°00.0'W	4240	3800	11/12/88 - 2/5/90	365	OP6	Wong et al $[1999]$
50°00.0'N	145°00.0'W	4240	1000	13/10/89 - 11/11/90	>360	OP7	Wong et. al. $[1999]$
50°00.0′N	145°00.0'W	4240	3800	13/10/89 - 11/11/90	>360	OP7	Wong et. al. [1999]
50°00.0′N	145°00.0′W	4240	3800	26/9/92-16/5/94	365	OP8	Wong et. al. [1999]
11°00.0′N	$140^{\circ}00.0'W$	4260	1600	29/12/82-14/2/84	412	S	Dymond and Collier [1988]
11°00.0′N	140°00.0′W	4260	3400	29/12/82-14/2/84	412	S	Dymond and Collier [1988]
48°59.8′N	173°54.8′W	5400	4806	11/8/91-27/6/92	321	SA2	Takahashi et al. [2000]
48°59.5′N	173°58.5′W	5427	4833	10/8/92-27/6/93	321	SA3	Takahashi et al. [2000]
49°00.9'N	17/4°00.0 W	5368	47/74	6/8/94-2/8/95	361	SA5	Takahashi et al. [2000]
4 28.0 N	8/ 19.0 E- 97°19.0/E	401/-	1040-	$\frac{28}{10/8} - \frac{28}{2/88}$	312	SBB1	Ittekkot et al. [1991]; Schäfer et al. [1996]
4°28 0'N	87°19.0′E	4045	3006-	$\frac{1}{4} \frac{30 - 0}{10} \frac{10}{30}$	312	SBB1	Ittekkot et al. [1991], Schäfer et al. [1996]
+ 20.0 IV	87°18.0′E	4045	2983	1/4/88 - 6/10/88	512	"	Ittekkot et al. [1991]; Schäfer et al. [1996]
5°01.0′N	87°09.0′E	3996	3010	5/12/90 - 26/10/91	325	SBB2	Ittekkot et al. [1991]; Schäfer et al. [1996]
32°05.0′N	64°15.0′W	4200	3200	7/2/80-10/12/80	293	SCF1	Deuser et al. [1981]; Conte et al. [2001]
32°05.0′N	64°15.0′W	4200	3200	6/2/81-3/2/82	358	SCF2	Deuser et al. [1981]; Conte et al. [2001]
32°05.0′N	64°15.0′W	4200	3200	18/1/83-17/2/84	364	SCF3	Deuser et al. [1981]; Conte et al. [2001]
0°00.2′N	175°09.7′E	4880	4363	1/6/92-16/4/93	319	SEC ^a	Kawahata et al. [2000]
78°53.4′N	6°44.5′E	1661	1110	01/07/88-04/06/89	338.5	SP2	Hebbeln [2000]
78°52.6′N	6°40.5′E	1676	1125	05/06/89-05/6/90	396	SP3	Hebbeln [2000]
16°18.0'N-	60°28.0'E	4018-	3021-	20/11/86-2/5/87	>305	WAS1"	Haake et al. [1993]; Ramaswamy and Nair [1994]
16°19.0'N-	"	4010-	3033-	12/5/8/-21/10/8/			Haake et al. [1993]; Ramaswamy and Nair [1994] Haake et al. [1992]; Ramaswamy and Nair [1994]
16°35.0'N	60°28 0'E-	4010	3039	22/11/87 - 31/10/88 22/11/87 - 31/10/88	>305	WAS2 ^a	Hacke et al. [1995], Ramaswamy and Nair [1994]
16°24 0'N	60°29.0′E	-010	3029	20/11/88 - 24/2/89	- 505	"	Hacke et al. [1993]; Ramaswamy and Nair [1994]
16°19.0′N	60°31.0′E	4013	3016	15/1/90 - 15/10/90	>305	WAS3 ^a	Haake et al. [1993]: Ramaswamy and Nair [1994]
2°59.8′N	135°1.5′E	4414	1592	4/6/91-15/4/92	317	WCB ^a	Kawahata et al. [1998]
2°59.8′N	135°1.5′E	4414	3902	4/6/91-15/4/92	317	WCB ^a	Kawahata et al. [1998]
20°04.2'S	9°10.0′E	2217	1640	4/3/88-16/3/89	376	WR1 ^a	Wefer and Fischer [1993]
20°02.8′S	9°09.3′E	2196	1648	18/3/89-13/3/90	405	WR2 ^a	Wefer and Fischer [1993]
20°03.0′S-	9°09.0′E-	2208-	1648	1/1/90-31/12/90	353	WR3 ^a	Fischer and Wefer [1996]
20°02.0'S	9°10.0′E	2263	1648		220	" 11/15 (⁹	Fischer and Wefer [1996]
20°02.0'S-	9°10.0 E-	2208-	1648-	1/1/91-1//12/91	339	WR4"	Fischer and Wefer [1996]
20 08.0 S	8 38.0 E	2263	1/1/				rischer and wejer [1990]

^a Trap experiments for which errors could be estimated.



Figure 2. Regression analysis between Particulate Organic Carbon (POC) and Particulate Organic Matter (POM) estimated from the combustible fraction and opal. Solid diamonds indicate sediment trap experiments. Open diamonds indicate water column data. The regression equation, correlation coefficient and probability are also given in the plot.

variability between data sets and the highest uncertainties (Table 2). These results could indicate that the fraction of carbon associated with the lithogenic fraction is sensitive to the sources and possibly mineralogy of lithogenic particles.

The analysis results for the selected trap data set shows the least variability in the carrying coefficients for lithogenic material and, therefore, seem to give the best results for open ocean conditions. In view of these results following analysis will, therefore, be limited to the selected trap data set.

[10] Carrying coefficients obtained using multiple linear correlation analysis are all significant. However, we find that a single-ballast model (following Armstrong et al. [2002]) is almost as good at predicting the overall variability of POC fluxes as the three-ballast model presented above (Table 3). These results seem to contradict the hypothesis that regional difference in POC flux and POC/ PIC rain ratios could be determined by changes in ballast composition. However, comparison of measured versus predicted values (Figure 4) indicates that the three-ballast model is better for predicting POC/PIC rain ratios. Consequently, the single ballast model overestimates POC fluxes in regions where opal dominates ballast fluxes such as the Bering Sea, the Subarctic Pacific and the Southern Ocean and underestimates POC fluxes in areas where CaCO₃ and lithogenic material dominate mineral fluxes. None of the models are very good at reproducing geographical variability of PIC/POC rain ratios in areas of intermediate to high CaCO3 versus opal and lithogenic material export fluxes (mass CaCO₃/total mass > 0.4, Figure 5), although the three-ballast model gives a better estimate of average values for both POC fluxes and POC/PIC rain ratios (Figure 5, Table 4).

[11] To investigate the causes for the additional regional variability of POC fluxes and POC/PIC rain ratios in areas of high relative CaCO₃ fluxes, we tested the possibility that the "free" POC fraction might still be a significant component of the fluxes. These tests assumed the presence of free POC, representing most of the export production near the surface, proportional to surface estimates of primary pro-



Figure 3. Scatterplots, correlation coefficient and probability of particulate organic carbon (POC) flux *versus* CaCO₃, opal and lithogenic material fluxes for annual trap experiments. Solid circles indicate trap experiments were the POC and opal fraction remineralized/dissolved in cups was added to the fluxes. Solid squares indicate trap experiments were the opal fraction dissolved in cups was added to the fluxes. Open circles indicate trap experiments not corrected for the particulate fraction remineralized/dissolved in the sampling cups. Error bars correspond to one standard deviation.



Figure 4. (a-f) Scatterplots of the measured versus predicted POC flux, and (g-l) measured versus predicted POC to CaCO₃ (as POC/PIC) rain ratio. Values were calculated for models considering 3 mineral fractions (Figures 4a–4c and 4g–4i) or considering only one mineral fraction (sum of total minerals) (Figures 4d–4f and 4j–4l) and using the selected trap experiments carried out below 1000 m depth (Figures 4a, 4d, 4g, and 4j), below 2000 m depth (Figures 4b, 4e, 4h, and 4k) and below 3000 m depth (Figures 4c, 4f, 4i, and 4l). Regression line (solid) and 1:1 line (dashed) between measured and predicted values is given. Solid circles indicate trap experiments were the POC and opal fraction remineralized/dissolved in cups was added to the fluxes. Solid squares indicate trap experiments were the opal fraction dissolved in cups was added to the fluxes. Open circles indicate trap experiments not corrected for the particulate fraction remineralized/dissolved in the sampling cups. Error bars correspond to 1 standard deviation.



Figure 5. Scatterplots of the ratio between predicted and measured values versus mass $CaCO_3$ /total mass ratio. Values are presented for models considering (a, c, e) three mineral fractions or considering (b, d, f) only one mineral fraction (sum of total minerals) and using the selected trap experiments carried out below 1000 m depth (Figures 5a and 5b), below 2000 m depth (Figures 5c and 5d) and below 3000 m depth (Figures 5e and 5f). The 1:1 line for the ratio between measured and predicted values is given. Solid circles indicate trap experiments were the POC and opal fraction remineralized/dissolved in cups was added to the fluxes. Solid squares indicate trap experiments not corrected for the particulate fraction remineralized/dissolved in the sampling cups.

	CaCO3	Opal	Lithogenic	r	N
All traps $< 1000 \text{ m}$	0.094 ± 0.010	0.025 ± 0.011	0.035 ± 0.006	0.962	107
Traps- $SP < 1000 \text{ m}$	0.090 ± 0.010	0.023 ± 0.011	0.052 ± 0.022	0.956	105
Traps $< 2000 \text{ m}$	0.075 ± 0.011	0.029 ± 0.009	0.052 ± 0.018	0.973	78
Traps < 3000 m	0.077 ± 0.012	0.030 ± 0.010	0.039 ± 0.035	0.970	50
Selected < 1000 m	0.083 ± 0.011	0.023 ± 0.010	0.068 ± 0.028	0.976	46
Selected $< 2000 \text{ m}$	0.074 ± 0.010	0.025 ± 0.008	0.071 ± 0.030	0.985	34
Selected $< 3000 \text{ m}$	0.070 ± 0.025	0.026 ± 0.009	0.065 ± 0.048	0.985	20

Table 2. Carrying and Correlation Coefficients for Multiple Correlation Analysis of POC Fluxes Versus Mineral Fluxes^a

^a Results are given for models separating the mineral fractions in CaCO₃, opal and lithogenic material and using trap experiments carried out below 1000 m (<1000 m), below 2000 m (<2000 m) and below 3000 m (<3000 m). All traps (all trap experiments compiled), traps-SP (all traps save SP*). Selected: traps with little or no influence of exogenous material; r: correlation coefficient; N: number of experiments used for the analysis. All correlations coefficients between measured and predicted values are significant (P< 0.0001). SP* traps were separated in the analysis because of the high contribution of ice-rafted debris [*Hebbeln*, 2000]. Errors correspond to the 95% confidence interval of the coefficients.

ductivity (PP) [Antia et al., 2001]. Primary productivity was derived from estimates by Antoine et al. [1996] and Behrenfeld and Falkowski [1997]. The decay scale for free POC was adopted from Armstrong et al. [2002] (e-folding scale of 560 m):

$$\mathbf{F}(\mathbf{z}) = f_{o}.\mathbf{F}_{o}(\mathbf{z}) + f_{ca}.\mathbf{F}_{ca}(\mathbf{z}) + f_{1}.\mathbf{F}_{1}(\mathbf{z}) + \beta.\mathbf{PP.e}^{(\Delta \mathbf{z}/\partial \mathbf{e})}$$
(4)

alternatively,

$$F(z) = f_{o}.F_{o}(z)f_{ca}.F_{ca}(z) + f_{1}.F_{1}(z) + \beta.PP/z$$
(5)

As for the carrying coefficients, the proportionality factor (β) determining free POC from primary productivity data can be solved through multiple regression analysis.

[12] Multiple correlation analysis of the selected trap data set indicates that the primary productivity term was not significant for trap experiments below 2000 m depth. When analyzing the selected data set with trap experiments below 1000 m depth, the free POC coefficient is significant only for the model assuming an inverse relationship between free POC and depth (β .PP/z). The improved model explains at most 22% of POC/PIC rain ratio variability for traps with high mass CaCO3/total mass ratios (still not very good). This corresponds to an total increase from 80 to 84% of the total variance in POC/PIC rain ratios explained for trap experiments below 1000 m depth, not better than results obtained using trap experiment below 2000 m and 3000 m depth (Table 3). In short, including a free POC fraction in the analysis reduces differences between results using data collected at different depth intervals, rather than improving significantly the predictions for POC/PIC rain ratios at high mass CaCO₃/total mass ratios. The same tests were done using only trap data with mass $CaCO_3/total$ mass >0.4, based on the assumption that in systems with high mass CaCO₃/total mass export flux the carrying coefficients for the different mineral fractions might differ. The results are similar to the previous analysis: variability in POC/PIC rain ratios cannot be explained by the three fractions ballast model independent of the depth range of the trap experiments used. In addition, the derived equations assuming a residual free POC fraction improved POC/ PIC rain ratio predictability by at most 24%, not better than models including trap data with low mass $CaCO_3/$ total mass ratios (<0.4).

[13] We also tested the possibility that the ballasted POC decays with depth, rather than sinking conservatively. In order to quantify the depth effect we compared the ratio between predicted values and measured values to depth:

$$F(z) = \{f_{o}.F_{o}(z) + f_{ca}F_{ca}(z) + f_{1}.F_{1}(z)\}/(\beta z)$$
(6)

[14] Results indicate a significant relationship with depth for trap data independent of the depth interval considered for the analysis. Comparison of predicted versus measured values indicates that some of the variability in the POC/PIC rain ratios from trap experiments with high mass CaCO₃/ total mass ratios can be explained by ballasted POC degradation at depth (Table 5) depending on trap data used. Including a depth dependent remineralization of ballasted organic carbon does not, however, improve predictions of POC fluxes and POC/PIC rain ratios when trap experiments with low mass CaCo₃/total mass ratios are included in the analysis (Tables 3, 4 and 5).

[15] In regions of high mass CaCO₃/total mass export flux, most of the variability in the POC/PIC rain ratio in the deep sea is not predictable by any version of the ballast model that we have been able to devise and is not related to differences in primary productivity (direct comparison of POC/PIC rain ratios in areas of high mass CaCO₃/total mass ratios with levels of primary production

Table 3. Variance Explained (in %) by the Models for POC Fluxes and POC/PIC Rain Ratios^a

	Three	e Fractions	One Fraction		
	POC	POC/PIC	POC	POC/PIC	
<1000 m	84	80	67	78	
<2000 m	89	87	71	86	
<3000 m	87	89	75	95	

^aResults are given for the model separating the mineral fractions in CaCO₃, opal and lithogenic material (three fractions) and the models using only the sum of mineral fractions (one fraction). Analysis was done using the selected trap experiments carried out below 1000 m depth (<1000 m), below 2000 m depth (<2000 m) and below 3000 m depth (<3000 m). All correlations coefficients between measured and predicted values are significant (P < 0.0001).

	<1000 m	<2000 m	<3000 m
	Three-Fractions Model ^b		
Percent of POC variance explained	83	88	84
Percent of POC/PIC variance explained	Ns	Ns	Ns
Average POC/PIC	0.852 ± 0.101	0.790 ± 0.114	0.765 ± 0.104
	One-Fraction Model ^c		
Percent of POC variance explained	79	87	80
Percent of POC/PIC variance explained	Ns	Ns	Ns
Average POC/PIC	0.714 ± 0.110	0.653 ± 0.102	0.574 ± 0.098
	Measured ^d		
Average POC/PIC	0.814 ± 0.241	0.768 ± 0.197	0.719 ± 0.215

Table 4. Average POC/PIC Rain Ratios (±1 Standard Deviation) and Variance Explained (in %) by the Models for POC Fluxes and POC/PIC Rain Ratios^a

^a Results are given for trap experiments with mass CaCO₃/total mass ratios above 0.4. Models were derived from the analysis of selected trap experiments carried out below 1000 m depth (<1000 m), below 2000 m depth (<2000 m) and below 3000 m depth (<3000 m).

^bModel based on three mineral fractions: CaCO₃, opal, lithogenic material (three fractions).

^cModel based on the sum of minerals (One Fraction).

^dAverage POC/PIC rain ratios calculated from the trap data (Measured).

is significant but corresponds only to at most 6% of the variability).

[16] The importance of $CaCO_3$ in transporting carbon to the deep ocean contradicts the hypothesis based on plankton dynamics and seasonal sediment trap flux showing high opal versus CaCO₃ ratios during periods of high POC flux [Ittekkot et al., 1991; Honjo, 1996; Takahashi et al., 2000]. We, therefore, collected individual cup measurements from the traps used in the analysis of annual data (Figure 1) to investigate variability of POC fluxes as a function of composition in the mineral fraction on short timescales (4-61 days). Correlation of POC fluxes with the different mineral fractions shows similar results as for annual data (Figure 6). Again, the highest correlation coefficient is obtained for CaCO₃. POC versus opal fluxes show variations dependent on the mass ratio of opal/total minerals. Applying the models obtained from the analysis of annual data to the cup data gives good results (for all trap data save SP* r = 0.899 and r = 0.902, and for the reduced data set r =0.909 and r = 0.982; for POC and POC/PIC respectively). These results confirm that the carrying coefficient as determined for the annual flux data are not an artifact of averaging but reflect mechanisms affecting POC fluxes both on an annual scale as well as on the scale relevant to biological and chemical processes.

4. Discussion

4.1. Predictability of POC Transport at Depth

[17] Multiple correlation analysis of deep-sea flux data from sediment trap experiments indicates that geographical variability in organic carbon fluxes can be accounted for, to a large extent, by the composition of associated sinking mineral particles, consistent with Armstrong et al. [2002]. Within a range of annual mass CaCO₃/total mass ratios between 0.4 and 0.8, however, estimates of average POC fluxes and the resulting POC/PIC rain ratios are similar to the measured values, but variability is poorly predicted. Measurement of particulate dissolution in trap cups could affect results as compared to the model. Few studies report values for dissolved components in the cups which may depend on sampling strategy (sampling time, preservative).

Table 5. Variance Explained (in %) by the Models for POC Fluxes and POC/PIC Rain Ratios and Average POC/PIC Rain Ratio (± 1 Standard Deviation)^a

	<1000 m	<2000 m	<3000 m
	All Trap Data		
Percent of POC variance explained	74	78	86
Percent of POC/PIC variance explained	36	72	87
	$CaCO_3 > 0.4$		
Percent of POC variance explained	59	78	88
Percent of POC/PIC variance explained	12	16	33
Average predicted POC/PIC ^b	1.019 ± 0.414	0.852 ± 0.203	0.738 ± 0.114
Average measured POC/PIC ^c	0.814 ± 0.241	0.768 ± 0.197	0.719 ± 0.215

^aResults are given for models derived using selected trap experiments below 1000 m (<1000 m), 2000 m (<2000 m) and 3000 m (<3000 m) depth and assuming that the mineral associated POC is also remineralized with depth. Results were obtained through linear regression of measured versus predicted values from trap experiments used to derive the models (all trap data) and a subset of the data with mass $CaCO_3/total mass ratios > 0.4$ ($CaCO_3 > 0.4$). All correlation coefficients between measured and predicted values are significant (P < 0.0001). ^bModel derived average POC/PIC rain ratios for the subset of trap data with mass CaCO₃/total mass ratios >0.4. Values for

all trap data are given in Table 4.

^c Average POC/PIC rain ratios calculated for the subset of trap data with mass CaCO₃/total mass ratios >0.4. Values for all trap data are given in Table 4.



Figure 6. Scatterplots, correlation coefficient and probability for individual cup measurements of POC flux versus $CaCO_3$, opal and lithogenic material fluxes for trap experiments below 1000 m depth. Slopes of the regression lines are 0.126 for $CaCO_3$, 0.061 for opal and 0.180 for lithogenic material. Shaded circles in the plot of POC versus opal represent the variations in mass opal/total mass ratios levels. The lighter shading corresponds to high mass opal/total mass ratios, dark shading to low mass opal/total mass ratios.

Values reported in the literature are highly variable ranging from 1.8 to over 30% for opal and 10% for POC [von Bodungen et al., 1991; Fischer and Wefer, 1991; Honjo and Manganini, 1993; Honjo et al., 1995]. The effect on annual estimates does not seem to exceed 10% as high dissolution tends to occur only during periods of low flux. Also, the scatterplots of POC *versus* other flux components (Figure 3) and predicted *versus* measured POC (Figure 5) do not reveal a systematic bias in trap experiments that were not corrected for dissolution effects. For several trap experiments we



Figure 7. Sinking speeds of model aggregates as a function of opal/CaCO_3 ratio (given as Si/Ca molar in the x axis) for mass POC/total mass ratios values of 0.4, 0.2, 0.1, 0.05, 0.025 and 0.01. The fraction of lithogenic material was assumed to be 9% of total mass, the average value in open ocean trap experiments. Sinking speeds are given for (a) a 100 μ m diameter aggregate and (b) a 1 mm diameter aggregate. The horizontal double arrows indicates the mass POC/total mass ratio of particles ballasted only by CaCO₃ (left) or opal (right) and having same sinking speed. The shaded area corresponds to the range of sinking speeds expected from the composition (mass POC/total mass ratio) of surface trap experiments (at 200 m depth). The framed area corresponds to the range of sinking speeds expected from the composition (mass POC/total mass ratio) of deep trap experiments (below 1000 m depth).

Table 6. Predicted Benthic Fluxes Obtained From Model Run [*Archer et al.*, 2000] and Benthic POC Flux Estimates Based on Carrying Coefficients Determined in This Study and Benthic Fluxes of CaCO₃, SiO₂ (Opal) and Clay

	Benthic flux	Predicted Org.C. carrying flux,			
	$10^{12} \text{ g yr}^{-1}$	$10^{12} \text{ g yr}^{-1}$	(% of total)		
Org. C ^a	886				
CaCO ₃	7438	521-617	(83-80%)		
SiO ₂	3255	84-95	(11 - 15%)		
Clay	600	39-41	(5-6%)		
Total		654-742			

^aOrganic carbon.

could calculate an estimate of prediction errors by propagation of analytical and splitting errors of the measurements. Error estimates for our predictions vary from ± 9 to $\pm 28\%$ for POC fluxes and from ± 10 to $\pm 35\%$ for POC/PIC ratios. Such estimates do not include the effects of trapping efficiency and dissolution in cups and indicate there is scope for improvement both in sediment trap methodology and reporting of results. Differences between predicted and measured values range from -59 to +129% of the measured values. Average values are $\pm 35\%$ for traps with annual mass CaCO₃/total mass ratios between 0.4 and 0.8 and $\pm 19\%$ for the other traps. Although our average error in estimates are close to what would be expected due to sample processing, the range observed far exceeds these values. Estimates of variability due to POC remineralization in deep trap experiments did not explain the differences between measured and predicted values. We cannot, however, totally exclude deep sea remineralization as causing at least part of the observed variability, given the relatively larger error on our predictions. Also, processes influencing remineralization at depth such as grazing and particle dynamics [Berelson, 2002] might be more localized and might not be resolved by an analysis on the global scale.

4.2. Theoretical Effect of Particle Composition on Sinking Speed

[18] Values for the carrying coefficients of mineral particles follow the order $CaCO_3$, lithogenic material, then opal. One potential explanation for the differences in carrying coefficient is the ballast effect. The sequence of the carrying coefficients varies in the same order as the mineral density: calcium carbonate (2.71 g cm⁻³), lithogenic material (e.g., quartz is 2.65 g cm⁻³), and opal (2.1 g cm⁻³). In order to test the differential effect of mineral on density, we estimated sinking speed of aggregates as a function of composition.

[19] Sinking rates of particle 100 μ m and 1 mm in diameter were calculated using the terminal velocity equation for a solid sphere at low Reynolds numbers (<0.5) and the empirical formula of *Zahm* [1927] at high Reynolds numbers [*Vogeler and Wolf-Gladrow*, 1993]. We assumed a density for organic matter of 1.06 g cm⁻³ (estimated density of a bacterial cell containing 80% water, *Logan and Hunt* [1987]). Porosity of particles was calculated using the relationship of *Alldredge and Gotschalk* [1988] between size and porosity. We used the empirical formula of *Matsumoto and Suganuma* [1977] to estimate the effect of particle

permeability on sinking rates assuming fibers diameter of 1 μm and 10 μm . Our estimates indicate that permeability increases sinking rate by 10% at most [Masliyah and Polikar, 1980] and should therefore not significantly affect our results. Theoretical estimates of particle sinking rates are presented as a function particle composition (ratio of opal/CaCO₃ presented as Si/Ca molar ratios) and mass POC/total mass ratios (Figure 7). The influence of Si/Ca on sinking rates is highest within the range found in most trap measurements (0.2 to 20). Sensitivity of sinking rates to changes in Si/Ca depends on mass POC/total mass and is not significant at values above 0.2. These calculations show that the density contrast between CaCO₃ and opal might be responsible for the observed factor of 2 difference in carrying coefficients. When the organic carbon fraction is 0.1 or below, variations in Si/Ca can affect the sinking velocity by up to 48% (Figure 7). This is similar to the difference in carrying capacity observed for Ca and Si when expressed in molar ratios (46%). Alternatively, at comparable sinking speeds particles containing only CaCO3 as ballast carry about 2 to 4 times as much POC than particles containing only opal as ballast (double arrows in Figure 7). This is similar to the difference in carrying coefficients based on mass between CaCO₃ and opal.

[20] It is unclear why sinking speeds should affect POC content of particles in the deep sea: our carrying coefficients were determined from analysis of trap experiments at depths ranging from 1000 to 4833 m and did not seem to be significantly affected by depth. In addition, bacterial activities measured in the water column and in sinking organic matter below 1000 m depth tend to be very low [Turley, 1993; Patching and Eardly, 1997]. In shallow waters, where a greater fraction of sinking POC is remineralized, differences in sinking speed might influence the quantities of carbon reaching the deep sea. Within the range of mass POC/total mass ratio observed in shallower traps (0.03 to 0.43 with annual averages ranging from 0.042 to 0.24 [Lohrenz et al., 1992; Wefer and Fischer, 1993; Pilskaln et al., 1996; Thunell, 1998a, 1998b; Peña et al., 1999; Collier et al., 2000; Conte et al., 2001] the ballast composition could also significantly affect sinking rates and, therefore, POC remineralization as a function of depth. As particle sinking speeds depend on a range of other parameters [Berelson, 2002], the effect of ballast composition on particle sinking rates and POC transport at depth is still speculative but warrants further investigation.

4.3. Implications of the Organic Carbon–Mineral Flux Association

[21] An implication of the ballast model is that changes in the sinking flux of opal [*Harrison*, 2000] or dust [*Petit et al*, 1999] might change the ratio of organic to inorganic carbon (the rain ratio). The rain affects the pCO₂ of the atmosphere by two mechanisms; by rearranging the alkalinity and total CO₂ in the water column, and by altering the global burial rate of CaCO₃, which in the long term drives changes in ocean pH and atmospheric pCO₂ [*Archer and Maier-Reimer*, 1994; *Archer et al.*, 2000]. *Archer and Maier-Reimer* [1994] conducted an experiment similar to the potential effect of ballast model on the rain ratio by altering the fraction of organic carbon degradation in the water column, while maintaining everything else the same. This experiment minimizes the effect of water column chemistry rearrangement, and forces atmospheric pCO₂ entirely from the seafloor. They found that an increase in organic rain to the seafloor of 33% (from 15 to 20 μ mol C cm⁻² yr⁻¹ global average) was sufficient to draw down pCO₂ by approximately 60 ppm. Their model did not include SiO₂, but a subsequent revision of the model did [Archer et al., 2000]. Global benthic fluxes of organic carbon, CaCO₃, biogenic SiO₂, and terrigenous material from this model are given in Table 6. These were tuned to reproduce the observed benthic chemistry and global burial rates. The model organic carbon benthic flux is gratifyingly close to the predictions of the ballast model using the model benthic fluxes of CaCO₃, biogenic SiO₂, and terrigenous material and the carrying coefficients listed in Table 2.

[22] The telling result is that globally, 83% of the organic carbon flux to the seafloor is accounted for by CaCO₃ ballast. Therefore, increasing the organic carbon rain by 33% without changing CaCO₃ fluxes would require roughly a tripling of both opal and clay rain rates. Dust fluxes were higher during glacial time by a factor of 2–3 globally [*Rea*, 1994; *Mahowald et al.*, 1999]; by itself, this would increase organic carbon flux to the seafloor by 5–10%, significantly but not enough to explain the entire glacial/interglacial pCO₂ change.

5. Summary

[23] The model of Armstrong et al. [2002] proposing an association of deep sea POC fluxes with mineral ballast was tested using a global compilation of deep sediment trap experiments and distinguishing three forms of mineral ballast: calcium carbonate, opal, and lithogenic material. Our results indicate that (1) organic carbon fluxes in the deep sea can be estimated from the fluxes of calcium carbonate, opal, and lithogenic material; (2) the carrying coefficients (or mineral-associated organic carbon fractions) are similar for calcium carbonate and lithogenic material, and about 3 times higher than the carrying coefficient for opal; (3) comparison of POC fluxes estimated using our carrying coefficients and global estimates of mineral fluxes derived from model studies are similar to model estimates of POC benthic fluxes; up to 83% of global POC fluxes are associated to calcium carbonate because it is denser than opal and more abundant than terrigenous material; and (4) our results argue against large changes in the organic carbon to calcium carbonate rain ratios in the past.

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